# Internuclear separations using least squares and neural networks for 46 new maingroup diatomics 

Ray Hefferlin<br>Southern Adventist University, Collegedale, TN 37315, USA


#### Abstract

Combined least-squares and neural-network forecasts for internuclear separations of main-group diatomic molecules, most with from 9 to 12 atomic valence electrons, are presented. We require that the standarddeviation bounds of the forecasts overlap each other; this requirement is met by 65 molecules, of which 46 seem not to have been studied previously. The composite errors average $0.1036 \AA$ on either side of the composite predictions. There is agreement with 33 of 41 independent test data (80.5\%), and those not in agreement fall outside the composite error limits by an average of $1.83 \%$.


## 1. Introduction

It is not the purpose of this work to achieve the precision now achieved by computation of experimental measurement, but to make some reasonably precise and accurate predictions from a periodic system of diatomic molecules. There is evidence for the hypothesis that individual atoms of diatomic molecules echo the periodic law: when the atomic number of either atom passes through an atomic magic number, the data have an extremum; and between successive extrema, main-group data vary similarly [1-3]. This paper reports predictions for spectroscopic constants to single-digit accuracy obtained by using this hypothesis, and thereby also tests the hypothesis. The presuppositions are that

1. the optimal independent variables are the period and group numbers of the atoms (R1, C1, R2, C2) - e.g., giving SiO the address (3, 4, 2, 6) - as shown in Fig. 1;
2. portions of the data space are smooth enough to support least-squares and neuralnetwork forecasts of single-digit precision and accuracy (Fig. 1);
3. least-squares and neural-network forecasts with overlapped error measures can be more precise and accurate than either alone.


Fig. 1. $r_{e}$ for diatomic molecules ranging from unipositive hydrides to uninegative period-3 species, plotted on the atomic-number space $Z_{1}, Z_{2}$. A vertical plane on the left-right diagonal passes through homonuclear molecules; the space is symmetric toward and away from the reader with respect to this plane. The ridges pertain to molecules with a raregas atom (helium, neon, and argon). The isolated peaks are due to alkaline-earth pairs ( $\mathrm{Be}_{2}$ and $\mathrm{Mg}_{2}$, and by presumption BeMg and $\mathrm{MgBe})$. Data points are all from [4] except for those taken from the literature to define the flanks of the ridges and the neighborhoods of the small peaks. This behavior continues out to period-6 molecules, so the space $\left(Z_{1}, Z_{2}\right)$ is cut into subspaces by high ridges bounding square shallow plains. The plains are enumerated by $R_{1}$ and $R_{2}$, which run from 2 to 6 . Within each plain, individual molecules are enumerated by $C_{1}$ and $C_{2}$; these group numbers run from 1 to 8 because only maingroup molecules are under investigation. Reproduced from [3] with permission from Edwin Mellen Press, Lewinston, New York.

The procedure consists of constructing a least-squares (LS) formula using internuclear-separation ( $r_{e}$ ) data from the 2009 edition of the Handbook of Physics and Chemistry [5]; and training neural network (NN) models on $r_{e}$ data from Huber and Herzberg’s 1979 compilation [4]; forecasting new data, along with error measures, using each method; combining the forecasts and error measures; and testing the combined results with independent literature results. This procedure is similar to that in [6].

Results for 65 molecules whose LS and NN standard deviations overlap are combined, and the average precision is $5.14 \%$, or $0.1036 \AA$ when absolute values are averaged. In a test of the accuracy of the predictions, 41 data values, for 22 molecules not in the compilations, were found in the literature; 33 (80.4\%) of the data fall within $0.1036 \AA$ of the tabulated value and the remainder fall outside by a very few percent.

## 2. The least-squares analysis

The data in the slightly concave plains of Fig. 1 were fitted with a quadratic polynomial in accordance with presuppositions 1 and 2 . For the plain $\left(R_{1}, R_{2}\right)=(2,2)$,

$$
\begin{align*}
r_{e}=3.53473 & -0.474877 C_{1}+0.031992 C_{1}^{2} \\
& -0.474877 C_{2}+0.031992 C_{2}^{2}+0.0297807 C_{1} C_{2} . \tag{1}
\end{align*}
$$

This polynomial must be normalized for other values of $\left(R_{1}, R_{2}\right)$ because the internuclear separations increase with period numbers. Nalewajski and Thakkar [7] show semiempirically that the normalization should depend on $\log \left(R_{1} R_{2}\right)$.

The description of procedures in this paragraph closely follows the more detailed description in [6]. Records for main-group molecules were sorted non-redundantly by $R_{1}$, $R_{2}, C_{1}$, and $C_{2}$. Main-group molecules with atoms from periods 2 to 6 were included, Eq. (1) was used to compute $r_{e}$ for each set of molecules with fixed period numbers. Good correlations with [4] were obtained for $\left(R_{1}, R_{2}\right)=(2,2)$ through $(2,6)$, $(3,3)$ through $(3,5)$, $(4,4)$ and $(4,5)$, and $(5,5)$. Within these combinations, when molecules with $\left(C_{1}, C_{2}\right)=$ $(1,1),(2,2)$, and $(7,7)$ were excluded the correlations improved markedly. After these selections, the predictions in each $\left(R_{1}, R_{2}\right)$ were divided by the slope of their trend line, thus producing the normalized predictions, hereafter denoted as $p$. The reciprocals of the original trend-line slopes define the normalization function $f\left(R_{1}, R_{2}\right)$, Eq. (2) and Fig. 2.

$$
\begin{equation*}
r_{e}=1.265-0.259 \ln \left(R_{1} R_{2}\right) . \tag{2}
\end{equation*}
$$



Product of the two atom period numbers

Fig. 2. The behaviors of the slopes of the trend lines of the original $r_{e}$ predictions, which are used to obtain normalized predictions.

Table 1. Distribution of percent errors, averaged over all $R_{1}$ and $R_{2}$, the on $C_{1}$ and $C_{2}$ plane

| 7 | $\begin{gathered} 2.576 \pm 2.994 \\ 15^{\mathrm{a}} \end{gathered}$ | $\begin{gathered} -0.425 \pm 2.155 \\ 10 \end{gathered}$ | $\begin{gathered} -2.679 \pm 2.429 \\ 15 \end{gathered}$ | $\begin{gathered} -3.707 \pm 2.234 \\ 7 \end{gathered}$ | $\begin{gathered} -1.063 \pm 2.657 \\ 6 \end{gathered}$ | $\begin{gathered} 7.994 \pm 1.153 \\ 4 \end{gathered}$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 6 | $\begin{gathered} 0.893 \pm 5.663 \\ 4 \end{gathered}$ | $\begin{gathered} 4.535 \pm 3.287 \\ 7 \end{gathered}$ | $\begin{gathered} 0.064 \pm 2.323 \\ 4 \end{gathered}$ | $\begin{gathered} -1.306 \pm 2.380 \\ 14 \end{gathered}$ | $\begin{gathered} -0.693 \pm 2.531 \\ 5 \end{gathered}$ | $\begin{gathered} 2.725 \pm 2.750 \\ 6 \end{gathered}$ |  |
| 5 |  |  | $\begin{gathered} -0.042 \pm 0 \\ 1 \end{gathered}$ | $\begin{gathered} -2.514 \pm 2.623 \\ 3 \end{gathered}$ | $\begin{gathered} -0.480 \pm 2.865 \\ 4 \end{gathered}$ |  |  |
| 4 |  |  |  | $\begin{gathered} -4.716 \pm 6.299 \\ 2 \end{gathered}$ |  |  |  |
| 3 |  |  | $\begin{gathered} -5.982 \pm 0 \\ 1 \end{gathered}$ |  |  |  |  |
| 2 |  |  |  |  |  |  |  |
| 1 |  |  |  |  |  |  |  |
|  | 1 | 2 | 3 | 4 | 5 | 6 | 7 |

> ${ }^{\text {a }}$ All entries at lower right of the major diagonal have been combined with, and placed at the locations of, the entries at upper left; there are no entries in column 7, which appears as incomplete.

The normalized data were then resorted according to $C_{1}$ and $C_{2}$; for each such group-pair the average differences between $p$ and data from [4], and their standard deviations ( $\sigma_{p}$ ), were found and averaged. A separate investigation showed that these measures had no
significant $\left(R_{1}, R_{2}\right)$ dependence. Table 1 presents the results. The space within the heavy line includes those addresses which have

1. unsigned percent errors of $5 \%$ or less
2. $\sigma_{p}$ of $7 \%$ or less
3. $p$ " $\sigma_{p}$ including zero, or a cohort of at least seven data $[(2,6),(3,7)$, and $(4,7)]$

The data were resorted according to $R_{1}, R_{2}, C_{1}$, and $C_{2}$. They consist of tabulated data from [4]; normalized LS predictions $p$; $\sigma_{p}$; adjusted normalized predictions $p$ ' and their standard deviation bounds $p^{\prime}-\sigma_{p^{\prime}}$ and $p^{\prime}+\sigma_{p^{\prime}}$. The last three quantities are illustrated by considering the group pair $(3,7)$ in Table 1: the -2.679 indicates that all of the normalized predictions included in this cell should be raised by $+2.679 \%$ and that the $\sigma_{p}$, limits then become $0.250 \%$ and $5.108 \%$.

## 3. Inclusion of the neural-network forecasts

The neural-network analysis [8] used data from [4] for training, validation, and testing; several trials (one described in Section 6 of [6]) indicate that the data from [4] and [5] are extremely well correlated. Records for all molecules that are in the final LS results were drawn from the table of NN results which are archived in the supplementary material to [8] (available from the American Chemical Society), and added to the LS file. The additions are the neural-network prediction $q$ and its absolute standard deviation bounds $q-\sigma_{q}$ and $q+\sigma_{q}$; the composite (Gaussian average) prediction $r_{e}$ (composite) and its composite error $\Delta r_{e}$ (composite). The average of this last quantity, which is the random error of the composite prediction process, is $0.136 \%$.

In accordance with the third presupposition stated above, it was determined to keep only those molecular records for which the $\sigma_{p}$, bounds overlap the $\sigma_{\mathrm{q}}$ bounds or vice versa. For example, the LS bounds for LiSe are $2.062 \AA$ and $2.310 \AA$, the NN bounds are $1.912 \AA$ and $2.159 \AA$, so there is an overlap. Keeping only these molecules reduces the number of records to 65 , with an average $\Delta r_{e}$ (composite) $0.1036 \AA$. They are listed in Table 2.

Table 2
Internuclear separation values and errors for molecules with overlapped LS and NN standard deviations, sorted alphabetically

| Molecule | R1 | C1 | R2 | C2 | $r_{e}$ (comp) | $\Delta r_{e}(\text { comp })$ <br> Absolute | Lower <br> bound | Upper bound |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| AlSe | 3 | 3 | 4 | 6 | 2.148 | 0.063 | 2.085 | 2.211 |
| Alte | 3 | 3 | 5 | 6 | 2.286 | 0.078 | 2.208 | 2.364 |
| Asl | 4 | 5 | 5 | 7 | 2.510 | 0.067 | 2.443 | 2.577 |
| AsSb | 4 | 5 | 5 | 5 | 2.262 | 0.065 | 2.197 | 2.326 |
| AsSn | 4 | 5 | 5 | 4 | 2.413 | 0.089 | 2.324 | 2.503 |
| AsTe | 4 | 5 | 5 | 6 | 2.353 | 0.091 | 2.262 | 2.444 |
| BeAt | 2 | 2 | 6 | 7 | 2.187 | 0.089 | 2.097 | 2.276 |
| Bel | 2 | 2 | 5 | 7 | 2.101a | 0.077 | 2.024 | 2.178 |
| BeSe | 2 | 2 | 4 | 6 | 1.858 | 0.081 | 1.777 | 1.940 |
| BeTe | 2 | 2 | 5 | 6 | 2.017 | 0.116 | 1.901 | 2.133 |
| BPo | 2 | 3 | 6 | 6 | 1.909 | 0.056 | 1.853 | 1.965 |
| BrSb | 4 | 7 | 5 | 5 | 2.510 | 0.067 | 2.443 | 2.577 |
| CAs | 2 | 4 | 4 | 5 | 1.661 | 0.062 | 1.600 | 1.723 |
| CaTe | 4 | 2 | 5 | 6 | 2.637 | 0.110 | 2.526 | 2.747 |
| CBi | 2 | 4 | 6 | 5 | 1.854 | 0.056 | 1.798 | 1.910 |
| CGe | 2 | 4 | 4 | 4 | 1.742 | 0.177 | 1.566 | 1.919 |
| ClSb | 3 | 7 | 5 | 5 | 2.354 | 0.063 | 2.291 | 2.416 |
| ClSr | 3 | 7 | 5 | 2 | 2.594 | 0.071 | 2.523 | 2.665 |
| CPb | 2 | 4 | 6 | 4 | 1.958 | 0.169 | 1.789 | 2.127 |
| CSn | 2 | 4 | 5 | 4 | 1.853 | 0.199 | 1.654 | 2.052 |
| CTe | 2 | 4 | 5 | 6 | 1.760 | 0.051 | 1.708 | 1.811 |
| FBi | 2 | 7 | 6 | 5 | 1.957 | 0.052 | 1.905 | 2.009 |
| GaTe | 4 | 3 | 5 | 6 | 2.442 | 0.078 | 2.364 | 2.520 |
| GeSb | 4 | 4 | 5 | 5 | 2.414 | 0.089 | 2.324 | 2.503 |
| GeSn | 4 | 4 | 5 | 4 | 2.523 | 0.204 | 2.319 | 2.728 |
| KTe | 4 | 1 | 5 | 6 | 2.947 | 0.336 | 2.610 | 3.283 |
| LiSe | 2 | 1 | 4 | 6 | 2.111 | 0.199 | 1.912 | 2.310 |
| LiTe | 2 | 1 | 5 | 6 | 2.088 | 0.007 | 2.081 | 2.095 |
| MgTe | 3 | 2 | 5 | 6 | 2.501 | 0.132 | 2.369 | 2.633 |
| NaS | 3 | 1 | 3 | 6 | 2.588 | 0.147 | 2.442 | 2.735 |
| NaSe | 3 | 1 | 4 | 6 | 2.649 | 0.196 | 2.453 | 2.845 |
| NAt | 2 | 5 | 6 | 7 | 1.994 | 0.089 | 1.905 | 2.083 |
| NBi | 2 | 5 | 6 | 5 | 1.950 | 0.051 | 1.900 | 2.001 |
| NGe | 2 | 5 | 4 | 4 | 1.673 | 0.050 | 1.623 | 1.723 |
| NI | 2 | 5 | 5 | 7 | 1.863 | 0.075 | 1.788 | 1.939 |
| NPb | 2 | 5 | 6 | 4 | 1.884 | 0.071 | 1.812 | 1.955 |
| NPo | 2 | 5 | 6 | 6 | 1.855 | 0.091 | 1.763 | 1.946 |


| NSb | 2 | 5 | 5 | 5 | 1.702 | 0.049 | 1.653 | 1.750 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| NSn | 2 | 5 | 5 | 4 | 1.783 | 0.060 | 1.724 | 1.843 |
| NTe | 2 | 5 | 5 | 6 | 1.721 | 0.069 | 1.652 | 1.790 |
| OTI | 2 | 6 | 6 | 3 | 1.893 | 0.072 | 1.821 | 1.965 |
| PCl | 3 | 5 | 3 | 7 | 2.091 c | 0.056 | 2.035 | 2.146 |
| PS | 3 | 5 | 3 | 6 | 1.933 | 0.049 | 1.884 | 1.982 |
| PSn | 3 | 5 | 5 | 4 | 2.254 | 0.074 | 2.180 | 2.328 |
| SAs | 3 | 6 | 4 | 5 | 2.059 | 0.100 | 1.960 | 2.159 |
| Sbl | 5 | 5 | 5 | 7 | 2.675 | 0.071 | 2.604 | 2.746 |
| SbTe | 5 | 5 | 5 | 6 | 2.460 | 0.091 | 2.369 | 2.551 |
| Seln | 4 | 6 | 5 | 3 | 2.432 | 0.089 | 2.343 | 2.520 |
| SeRb | 4 | 6 | 5 | 1 | 2.942 | 0.067 | 2.876 | 3.009 |
| SeSb | 4 | 6 | 5 | 5 | 2.361 | 0.099 | 2.262 | 2.459 |
| SeSr | 4 | 6 | 5 | 2 | 2.613 | 0.086 | 2.527 | 2.699 |
| SeTe | 4 | 6 | 5 | 6 | 2.355 | 0.091 | 2.264 | 2.446 |
| SGa | 3 | 6 | 4 | 3 | 2.144 | 0.059 | 2.085 | 2.203 |
| SiGe | 3 | 4 | 4 | 4 | $2.233 a$ | 0.140 | 2.083 | 2.363 |
| SIn | 3 | 6 | 5 | 3 | 2.272 | 0.092 | 2.180 | 2.364 |
| SiP | 3 | 4 | 3 | 5 | 1.992 | 0.071 | 1.920 | 2.063 |
| SiSb | 3 | 4 | 5 | 5 | 2.253 | 0.073 | 2.180 | 2.327 |
| SiSi | 3 | 4 | 3 | 4 | $2.082 a$ | 0.191 | 1.891 | 2.272 |
| SiSn | 3 | 4 | 5 | 4 | 2.358 | 0.200 | 2.157 | 2.558 |
| SiTe | 3 | 4 | 5 | 6 | 2.235 | 0.083 | 2.152 | 2.318 |
| SK | 3 | 6 | 4 | 1 | 2.637 | 0.208 | 2.429 | 2.845 |
| SnSb | 5 | 4 | 5 | 5 | 2.520 | 0.091 | 2.429 | 2.611 |
| SnSn | 5 | 4 | 5 | 4 | $2.623 a$ | 0.284 | 2.339 | 2.907 |
| SRb | 3 | 6 | 5 | 1 | 2.633 | 0.305 | 2.328 | 2.939 |
| SSb | 3 | 6 | 5 | 5 | 2.210 | 0.089 | 2.121 | 2.299 |

(a) Non-standard configuration [9]

## 4. Accuracy

Three tests of the composite predictions were performed to determine if the composite predictions are accurate:
a) Of the 65 composite predictions, there are data for 14 of them in [9]; of these, 10 data agree and four do not [they lie outside of $\Delta r_{e}$ (composite) by $5.27 \%, 2.33 \%$, $0.09 \%$, and $-1.64 \%$ ] respectively.
b) Of 24 data for six molecules given by [10], 21 agree and three do not [all three relate to one molecule and are $1.47 \%, 1.15 \%$, and $0.74 \%$ outside of $\Delta r_{e}($ composite $)$. One of the 21 in agreement has an anomalous aufbau [9].
c) Of three data from the new section of high-precision values in the Web site [4], two agree and one does not [ $2.17 \%$ outside of $\Delta r_{e}$ (composite)].

In sum, there is agreement with 33 of 41 test data (80.5\%), which is consistent with the conjecture that $\Delta r_{e}$ (composite) exceeds one "standard deviation." The agreements might have been even a larger percent of the 41 had the errors associated with the literature data been included.

## 5. Discussion

There are 46 composite predictions for which no tabulated or literature data were found. Do these contribute truly new information? The answer is not known because data for some of these molecules might not have been found in our searches of the literature. However, the last section suggests that there is an $80.5 \%$ chance that any new datum will agree with the combined predictions given here to within $\Delta r_{e}$ (composite) and that if it does not, then it will fall outside $\Delta r_{e}$ (composite) by a very few percent.

There are molecules with anomalous electron configurations; do they affect the results? Test (a) above: of the 10 molecules with composite predictions in agreement, two have anomalous aufbau; of the four in disagreement, one ( $0.09 \%$ outside) has anomalous aufbau [9]. Test (c): one of those in agreement has an anomalous aufbau [9]. It is concluded that having an anomalous configuration has not affected the LS and NN prediction process significantly.

Current experimental or computational predictions for $r_{e}$ are far more precise than those presented here and in some decades will replace them. In the meantime, it is hoped that the data given here might be useful as first approximations; that the joint use of LS and NN, common in mathematical chemistry, will have been shown useful in another area; and that the original hypothesis, as manifested in periodic systems of molecules using period and group axes [3, 11, 12], will have been further supported.

## Acknowledgements

The author thanks Professor A. I. Boldyrev (Utah State University) for an alert concerning non-aufbau electronic configurations; undergraduate student Ms. Amy Beard for her assistance in the work through part of Section 2; a reviewer for his justifiable criticisms, and the Southern Adventist University Endowed Fund for International Research in Physics.

## References

[1] Shchukarev S. A., Neorganicheskaya khimiya, Vol. 1, Vysshaya Shkola (Moscow), 1970.
[2] Kong F.-A., The Periodicity of Diatomic Molecules, J Mol Struct., 1982, 90, 17-28.
[3] Hefferlin R., Periodic systems of molecules and their relation to the systematic analysis of molecular data, Edwin Mellen Press, Lewiston, New York, 1989.
[4] Huber K. P., Herzberg G., Constants of Diatomic Molecules, Van Nostrand Reinhold, 1979. Data for individual molecules are available on-line, courtesy of the National Institute of Standards and Technology, by going to the site:
http://www.webbook.nist.gov/chemistry.
[5] Lide D. R., Editor in chief, Handbook of physics and chemistry. Taylor and Francis;2009.
[6] Hefferln R., Vibration frequencies using least squares and neural networks for 50 new $s$ and $p$ electron diatomics, J. Quant. Spectr. Radiat. Transf., 2010, 111, 71-77.
[7] Nalewajski R. F., Thakkar A. J., Correlations between average atomic numbers and spectroscopic constants of diatomic molecules, J. Phys. Chem., 1983, 87, 5361-5367.
[8] Hefferlin R., Davis W. B., An atlas of forecasted molecular data. 1. Internuclear separations of main-group and transition-metal neutral gas-phase diatomic molecules in the ground state, J. Chem. Inf. Mol. Model., 2006, 46, 820-825.
[9] Boldyrev A. I., Simons J., Periodic table of diatomic molecules, wall chart A, Wiley, 1997.
[10] Ruette F., Sanches M., Añez R., Bermúdez A., Sierraalta A., Diatomic molecule data for parametric methods, Int. J. Mol. Struct. THEOCHEM, 2005, 729, 19-37.
[11] Hefferlin R., Zhuvikin G., Caviness K., Duerksen P. J., Periodic systems of $N$-atom Molecules, J. Quant. Spectr. Radiat. Transf., 1984, 32, 257-268.
[12] Hefferlin R., Matrix-product periodic systems of molecules. JQRST 1994, 34, 314317.

