

The Periodic Table of the Stable Isotopes 1 - 25

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Abstract: In MHCE8S theory the stable isotopes are of special interest. Data from Wikipedia. 1st duplication sulfur. 1st abundant duplication zirconium

	stable isotopes	abundance(s)	duplications
1 hydrogen	1, 2		
2 helium	3, 4		
3 lithium	6, 7		
4 beryllium	5, 9		
5 boron	10, 11		
6 carbon	12, 13		
7 nitrogen	14, 15		
8 oxygen	16, 17, 18		
9 fluorine	19		
10 neon	20, 21, 22	90.4%, 0.27, 9.25	
11 sodium	23		
12 magnesium	24, 25, 26	79.0%, 10.0, 11.0,	
13 aluminium	27		
14 silicon	28, 29, 30	92.2%, 4.7, 3.1	
15 phosphorus	31		
16 sulfur	32, 33, 34, 36	94.99%, 0.75, 4.25, 0.01	
17 chlorine	35, 37		
18 argon	36 , 38, 40	0.33%, 0.06, 99.6	
19 potassium	39, 41		
20 calcium	40 , 42, 43, 44, 46	96.9%, 0.65, 0.135, 2.09, 0.004	
21 scandium	45		
22 titanium	46 , 47, 48, 49, 50	8.2%, 7.4, 73.7, 5.4, 5.18	
23 vanadium	51		
24 chromium	50 , 52, 53, 54	4.34%, 83.7, 9.50, 2.36	
25 manganese	55		