Summary
This paper further explores intuitions we highlighted in previous papers already:

1. The concept of the matter-wave traveling through the vacuum, an atomic lattice or any medium can be equated to the concept of an electric or electromagnetic signal traveling through the same medium.
2. There is no need to model the matter-wave as a wave packet: a single wave – with a precise frequency and a precise wavelength – will do.
3. If we do want to model the matter-wave as a wave packet rather than a single wave with a precisely defined frequency and wavelength, then the uncertainty in such wave packet reflects our own limited knowledge about the momentum and/or the velocity of the particle that we think we are representing. The uncertainty is, therefore, not inherent to Nature, but to our limited knowledge about the initial conditions.
4. The fact that such wave packets usually dissipate very rapidly, reflects that even our limited knowledge about initial conditions tends to become equally rapidly irrelevant. Indeed, as Feynman puts it, “the tiniest irregularities” tend to get magnified very quickly at the micro-scale.

In short, as Hendrik Antoon Lorentz noted a few months before his demise, there is no reason whatsoever “to elevate indeterminism to a philosophical principle.”

Contents
De Broglie’s wave packet: concepts and issues ......................................................................................... 1
De Broglie’s intuitions: right or wrong? ................................................................................................. 4
The Zitterbewegung or ring current model of an electron ...................................................................... 6
Schrödinger’s wave equation in free space ............................................................................................. 8
Schrödinger’s wave equation in a crystal lattice ..................................................................................... 12
  Feynman’s lattice .................................................................................................................................. 12
  The difference between a traveling electric charge and a traveling electric signal ................. 15
  The energy of an electron in a crystal lattice ....................................................................................... 17
One single matter-wave or a wave packet? ............................................................................................. 18
Conclusions ............................................................................................................................................. 21
Matter-waves, signals, electrons, and probability amplitudes

De Broglie’s wave packet: concepts and issues

Most models – including the model of Louis de Broglie himself\(^1\) – of de Broglie’s modelling of a matter-particle lead to the following issues in the interpretation of de Broglie’s theory of the matter-wave:

1. Heisenberg’s uncertainty principle, which assumes an *Ungenauigkeit*\(^2\) or an *Ungewissheit*\(^3\) in the energy value to be used in the (complex-valued) wavefunction which represents the matter-particle, leads physicists to consider a wave *packet* as a truthful representation of the matter-particle. A wave packet is a *superposition of component* waves of slightly different frequencies \(f_i = E_i/h\) and with slightly different wavelengths \(\lambda_i = h/p_i = h/m_i v_i\), as opposed to a single wave.

This epistemological or ontological predilection effectively amounts to considering uncertainty to be part of Nature or – as H.A. Lorentz famously put it – “elevating it to a philosophical principle” as opposed to relating it to (1) our lack of precise knowledge about initial conditions\(^4\) and/or (2) accepting the precision of any *measurement* will depend on the experimental set-up and/or the precision of the measurement apparatus.

**Note:** Apart from uncertainty as a epistemological principle or as the imprecision in a measurement, we may, perhaps, identify yet another interpretation of uncertainty: the detail or, its corollary, the lack of detail in our theoretical models. Such detail often evolves as a result of the precision of our measurements. An example that readily comes to mind here is the theoretical explanation for the hydrogen spectrum. Schrödinger’s wave equation for electron orbitals was, effectively, hailed as a truthful model\(^5\) – or more precise and, therefore, more correct than the Bohr-Rutherford model – because it was able to also explain the fine structure of the hydrogen spectrum. However, it is quite obvious that there is also a hyperfine structure due to the coupling between the magnetic moments of the electron and the proton in a hydrogen atom.\(^6\) Schrödinger’s model is, therefore, also incomplete and

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1 We consider Louis de Broglie’s paper for the 5th Solvay Conference (*La Nouvelle Dynamique des Quanta*, 1927) to be the original truthful representation of these views.

2 We prefer to use Heisenberg’s very first references to the concept of uncertainty to avoid the metaphysical connotation it acquired when, *paraphrasing* H.A. Lorentz, physicists decided to discard the assumption of causality by elevating indeterminism to a philosophical principle. Heisenberg used the term *Ungenauigkeit* to refer to a disturbance of the phenomenon by the measurement (in Compton scattering experiments, the classical velocity of an electron (both its direction as well as its magnitude) is changed as result of the probing by photons), which is related to the imprecision that is inherent to the measurement apparatus. A typical example of the latter is, effectively, the use of photons that have lower energy (and, therefore, longer wavelengths) to limit the impact of the measurement: as the wavelength of the photon determines the order of magnitude of the relevant length scale, a compromise must be made between the precision of the measurement and the disturbance it causes.

3 The term *Ungewissheit* is yet another term avoiding Heisenberg’s *Unbestimmtheit*. We note the Uncertainty Principle is usually referred to as the *Unbestimmtheitsprinzip* or the *Unschärferelation* in German (see, for example, the German-language Wikipedia article on *die Heisenbergsche Unschärferelation*).

4 If we knew the initial (base) state of the particle, we would not need the quantum-mechanical explanation: we would be able to explain the final state in terms of the initial state and classical physics explaining whatever happens between the initial and finale state.

5 It is often mentioned that the Uncertainty Principle explains why an electrons just does not go sit on top of a proton, but the Bohr-Rutherford model of an atom did the same based on the Planck-Einstein relation. The reader should also note here that there is no uncertainty whatsoever in Schrödinger’s wave equation for the bound electron: the energy levels of the subshells are what they are, and *exactly so*.

6 We think of the enigmatic Lamb shift as an even finer substructure which may or may not be explained by the anomaly in the magnetic moment of an electron which, as we argue in our paper on the electron structure, is not an anomaly at all: an easy
Feynman’s praise for it as “the most dramatic success in the history of the quantum mechanics” \(^7\) is, therefore, plain hyperbole: whatever it is that Schrödinger’s equation for bound electrons actually represents, it gives us the energy levels and geometry of atomic subshells only. We may also remind the reader it is hard, if not impossible, to use the model to explain multi-nucleon atoms, let alone the structure of the nucleus itself. Hence, any detail below that of an electron subshell is left out of the model.

2. Physicists then assume the dispersion relation that relates the \(f_i\) frequencies (or the angular frequencies \(\omega_i = 2\pi f_i\)) to the \(\lambda_i\) wavelengths (or the wavenumbers \(k_i = 2\pi / \lambda_i\)) should be directly related to the relativistic energy-moment relation\(^8\):

\[
E^2 - p^2c^2 = m^2c^4 \iff \frac{\hbar^2\omega^2}{\lambda^2} = \hbar^2\omega^2 - \frac{\hbar^2}{(2\pi / k)^2}c^2 \iff \hbar^2\omega^2 - \hbar^2k^2c^2 = m^2c^4
\]

This relation mixes two energy and/or mass concepts. One should effectively remember the following:

1. The \(m\) in the \(E^2 - p^2c^2 = m^2c^4\) relation is the rest mass and should, therefore, be written as \(m_0\).\(^9\)

2. The (linear) momentum is related to the kinetic energy of the particle only which only becomes a significant fraction of the total energy when the classical velocity of the particle starts approach the speed of light which – in most situations – it does not.

There is, of course, no reason to not assume the Planck-Einstein relation does not apply to the rest mass (on the contrary, it is generally considered to be as universal or as fundamental as Einstein’s mass-energy mass-equivalence relation \(E = mc^2\)) and we may, therefore, re-write the (dispersion) relation above as:

\[
\frac{\hbar^2\omega^2}{c^2} - \hbar^2k^2 = m_0c^2 = E_0 = \hbar\omega_0
\]

This, however, leads to a contradiction when assuming phase velocities equal to \(c = \lambda f = \omega/k\):

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\(^7\) See: Feynman’s Lectures on Quantum Mechanics, Chapter 19: The Hydrogen Atom and the Periodic Table.

\(^8\) We also have the \(pc = Ev/c = p = Ev/c^2\) relation, of course, but is rather significant one cannot derive this dispersion relation from it. When doing the same substitutions (\(p = \hbar k, E = mc^2\) and \(v = p/m\)), we just get the trivial \(\hbar k = \hbar k\) relation. The reader can also easily verify that the use of the Planck-Einstein relation \((E = \hbar\omega)\) leads to the same triviality:

\[
p = \hbar k = \frac{Ev}{c^2} = \frac{h\omega}{c} = \frac{\hbar\omega}{mc^2} = \frac{\hbar\omega k}{\hbar\omega} = \hbar k = p
\]

In fact, the very different result – an identity versus an inconsistency – should make us think about the (non)sense of the dispersion relation one gets out of the momentum-energy relation.

\(^9\) The reader will probably be aware of this but, if not, we refer him to any standard textbook. For ease of reference and because of its online availability, we used Feynman’s Lectures rather consistently over the past. The reader will find the derivation of the formula in section 5 of Chapter 16 of the first volume.
\[
\frac{h^2 \omega^2}{c^2} - \hbar^2 k^2 = h^2 \left( \frac{\omega^2}{c^2} - k^2 \right) = \frac{m_0 c^2}{c^2} = E_0 = h \omega_0 \iff h \left( \frac{\omega^2}{c^2} - k^2 \right) = \omega_0
\]

\[\Rightarrow h(k^2 - k^2) = h \cdot 0 = 0 \neq \omega_0\]

It is a contradiction because we cannot equate \(\omega_0\) to zero: the rest mass of our particle is not zero and, therefore, \(\omega_0\) cannot be zero. The only way out is to assume phase velocities are not equal to \(c\).\(^{10}\)

Indeed, if this dispersion relation would be correct, then we get superluminal phase velocities. You have, without any doubt, seen how this can be calculated. We can use the \(v_{\text{phase}} = \omega/k\) relation and the misguided substitutions for the momentum (or the wavenumber):

\[v_{\text{phase}} = \frac{\omega}{k} = \frac{E/h}{p/h} = \frac{m_0 c^2}{m \nu} = \frac{c^2}{\nu} = \frac{c}{\beta} \geq c\]

This is, in fact, what physicists – since Louis de Broglie first wrote this – have consistently been jotting down. We must admit we are flabbergasted by the ease with which physicists – including Albert Einstein, who seconded Louis de Broglie’s 1924 PhD thesis – accepted this inevitable result. Academic physicists usually justify this by noting these component waves cannot transmit any signal because the group velocity – the velocity of the wave packet – as a whole is all that actually matters, and this group velocity does effectively correspond to the classical velocity of the particle that the wave packet must, somehow, represent.\(^{11}\) Indeed, differentiating the dispersion relation and combining it with the erroneous relation for the phase velocity above yields what Louis de Broglie wanted to prove:

\[h^2 \omega^2 - \hbar^2 k^2 c^2 = m_0 c^4 \Rightarrow 2h^2 \omega \delta \omega = 2h^2 c^2 \delta k\]

\[\iff v_{\text{group}} = \frac{\partial \omega}{\partial k} = c^2 \frac{k}{\omega} = \frac{c^2}{v_{\text{phase}}} = \frac{c^2 \beta}{c} = v_{\text{particle}}\]

The inverse proportionality between the group and phase velocity is rather remarkable because, among other nonsensical results, it obviously implies infinite phase velocities for \(\nu = 0\). Another nonsensical result is this: while accepting the rather outrageous implication of superluminal phase velocities, we also accept a wave packet that cannot represent any real-life particle because it quickly dissipates. Indeed, the remarks above bring us to the third and final issue with mainstream interpretations of the matter-wave.

Before we will discuss that, however, we cannot help noting another small but significant inconsistency in de Broglie’s interpretation of the matter-wave: we already noted that mainstream theorists write the slightly diverging frequencies as \(f_i = E_i/h\) and the slightly diverging wavelengths as \(\lambda_i = h/p_i = h/m_i \nu\). This raises the following obvious question: why assume an uncertainty in the energy \(E\) and the mass \(m\) but

\(^{10}\) To get out of the contradiction, one might also assume that each of the component waves carries an infinitesimally small part of the energy (or, what amounts to the same, the equivalent mass) but, while one could work out a ‘mass without mass’ model of a particle based on such rather reasonable assumption, that is not how mainstream physicists proceed: they may assume some uncertainty about the energy or mass but they do not think in terms of the matter-wave actually carrying the energy or mass of the matter-particle.

\(^{11}\) They sometimes add we should not think of the matter-wave as a real wave anyway. However, this is nonsensical: why should we think of the matter-wave as unreal if we talk phase velocity but, then, when talking the group velocity, think of it as something that actually is oscillating and moving in space at a velocity that corresponds to the classical velocity of the particle it is supposed to represent?
not in the velocity \(v\): if we use a subscript for the \(E\) and \(m\) – effectively assuming there is no precise value for them – then plain logic suggests that we should not assume a precise value for the classical velocity either. Let us now introduce the final and most glaring issue with the mainstream representation of a matter-particle by a wave packet.

3. Prof. H. Pleijel, then Chairman of the Nobel Committee for Physics of the Royal Swedish Academy of Sciences, dutifully notes the following on the nature of the new ‘matter waves’ in the ceremonial speech for the 1933 Nobel Prize, which was awarded to Heisenberg for “the creation of quantum mechanics”

"Matter is formed or represented by a great number of this kind of waves which have somewhat different velocities of propagation and such phase that they combine at the point in question. Such a system of waves forms a crest which propagates itself with quite a different velocity from that of its component waves, this velocity being the so-called group velocity. Such a wave crest represents a material point which is thus either formed by it or connected with it, and is called a wave packet. [...] As a result of this theory on is forced to the conclusion to conceive of matter as not being durable, or that it can have definite extension in space. The waves, which form the matter, travel, in fact, with different velocity and must, therefore, sooner or later separate. Matter changes form and extent in space. The picture which has been created, of matter being composed of unchangeable particles, must be modified."

This sounds very familiar: it is the basics of the basics of the mainstream interpretation of de Broglie’s matter-wave, in fact. The problem is this: it is, obviously, untrue. Real-life particles – electrons or atoms traveling in space – do not dissipate. They might change form and extent in space when traveling through one or more slits but, as Feynman and all mainstream physicists dutifully note I the context of the double-slit experiment with electrons: have to admit: electrons – or matter-particles in general – always come in identical lumps at the backstop: all these lumps are the same size, only whole lumps arrive, and they arrive one at a time.

So what is wrong and what is right here?

De Broglie’s intuitions: right or wrong?

De Broglie’s intuition in regard to the wave nature of matter is, essentially, correct. However, his modeling of it as a wave packet is not: modeling matter-particles as a simple two-dimensional oscillation in space does the trick. It is, however, a rather special oscillation: we interpret the real and imaginary part of the elementary wavefunction as the sine and cosine components of the orbital motion of a pointlike charge. The pointlike charge and its motion combine to explain Schrödinger’s discovery of the Zitterbewegung of the electron, which Paul Dirac describes as follows in his Nobel Prize speech:

“...It is found that an electron which seems to us to be moving slowly, must actually have a very high frequency oscillatory motion of small amplitude superposed on the regular motion which

12 To be precise, Werner Heisenberg actually got a postponed prize from 1932: it is Erwin Schrödinger and Paul A.M. Dirac who, jointly, got the actual 1933 prize. Prof. Pleijel therefore acknowledges all three more or less equally in the introduction of his speech: “This year’s Nobel Prizes for Physics are dedicated to the new atomic physics. The prizes, which the Academy of Sciences has at its disposal, have namely been awarded to those men, Heisenberg, Schrödinger, and Dirac, who have created and developed the basic ideas of modern atomic physics.”

13 See: Richard Feynman, Vol. III, Chapter 1, Section 4, An experiment with electrons.
appears to us. As a result of this oscillatory motion, the velocity of the electron at any time equals the velocity of light. This is a prediction which cannot be directly verified by experiment, since the frequency of the oscillatory motion is so high and its amplitude is so small. But one must believe in this consequence of the theory, since other consequences of the theory which are inseparably bound up with this one, such as the law of scattering of light by an electron, are confirmed by experiment.” (Paul A.M. Dirac, Theory of Electrons and Positrons, Nobel Lecture, December 12, 1933)

The de Broglie relation \( (p = \hbar/\lambda = \hbar k) \) is correct too, but it has a precise geometric interpretation which one can only appreciate by noting that the linear momentum of a particle is a vector quantity and that we, therefore, should probably also think of Planck’s quantum as a vector quantity: \( \hbar \) has a magnitude as well as a direction. We should, likewise, think of the reduced Planck constant \( (\hbar = \hbar/2\pi) \) as a proper angular momentum, which can and should be written as \( \hbar = I \omega \): the product of an angular mass (the rotational inertia \( I \)) and an orbital angular frequency \( (\omega) \). This, then, also gives meaning to the concept of spin (which is either up or down). Finally, this oscillatory motion also generates a classical magnetic moment which – equally classically – will precess in an external electromagnetic field. There is no uncertainty in this model, except for the uncertainty in regard to the initial plane of oscillation (which is given by the direction of \( \hbar \) and \( \omega \)). We have explored this model ad nauseam elsewhere so we will limit ourselves to a very brief presentation of it below only before we proceed to the main topic of our paper, which revolves around sensible wave equations and how the matter-wave may or may not move through a medium: the vacuum itself or, in a more practical context, a atomic lattice, such as semiconductor material.

Indeed, we want this paper to be a first in a series focusing on the relevance of the ring current model of matter-particles for quantum computing and semiconductor engineering. Indeed, while we got some good feedback from other researchers – both academic as well as non-professional or amateur physicists (as we are) – we think the concept of a wave equation to model the properties of free space (or the vacuum as we know it) is rather limited: we think Maxwell’s equations model these properties quite well and we, therefore, do not see the need to add some new wave equation modeling the properties of the vacuum in the context of matter-waves: if an electromagnetic wave cannot travel any faster than light, why would the (components of the) matter-wave be able to do so?

In contrast, the phase or group velocity of a matter-wave in a crystal lattice, a conductor or semiconductor would, obviously, correspond to the velocity of the signal which may or may not reach lightspeed and, hence, our reflections in this regard may, therefore, have an impact of how we think of signals and the processing thereof.

Likewise, analyzing the uncertainty in regard to the spin of an electron (up or down) as an uncertainty in the actual direction of the actual magnetic moment of an electron may impact on how we think about the difficulties related to decoherence and quantum noise – and the related quantum error correcting (QEC) so as to improve fault-tolerance in quantum computing.

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14 This is Paul A.M. Dirac’s speech at the Nobel Banquet in Stockholm, December 10, 1933.

15 We have documented the rather gradual development of what we refer to as a fully-fledged realist interpretation on Phil Gibb’s alternative science site (https://vixra.org/author/jean_louis_van_belle).
Let us not get ahead of ourselves here. We will first present the basics of the Zitterbewegung or ring current model of an electron and then return to our discussion of wave equations.

The Zitterbewegung or ring current model of an electron

We request the reader to think of the (elementary) wavefunction \( r = \psi = a \cdot e^{\text{i} \theta} \) as representing the physical position of a pointlike elementary charge – pointlike but not dimensionless\(^{16}\) – moving at the speed of light around the center of its motion in a space that is defined by the electron’s Compton radius \( a = h/mc \). This radius – which effectively doubles up as the amplitude of the wavefunction – can easily be derived from (1) Einstein’s mass-energy equivalence relation, (2) the Planck-Einstein relation, and (3) the formula for a tangential velocity, as shown below:

\[
\begin{align*}
E &= mc^2 \Rightarrow mc^2 = \hbar \omega \\
E &= \hbar \omega \\
\therefore c &= a \omega \Leftrightarrow a &= \frac{c}{\omega} \Leftrightarrow \omega &= \frac{c}{a} \\
\Rightarrow ma^2 \omega^2 &= \hbar \omega \Rightarrow m \frac{c^2}{\omega^2} \omega^2 = \hbar \frac{c}{a} \Leftrightarrow a &= \frac{\hbar}{mc}.
\end{align*}
\]

The exceedingly simple derivation\(^{17}\) above gives us a geometric interpretation of Prof. Dr. Patrick R. LeClair’s understanding of the Compton wavelength as “the scale above which the particle can be localized in a particle-like sense.”\(^{18}\) It effectively amounts to a very basic ring current model\(^{19}\) which allows us to interpret the elementary wavefunction \( r = a \cdot e^{\text{i} \theta} = a \cdot e^{\text{i}(E \cdot t - \text{p} \cdot x)} \) as the position vector of the pointlike charge as Dirac’s pointlike electric charge.

This Zitterbewegung (or ring current) electron may now move linearly and we may conveniently and without any loss on the generality of the argument choose our \( x \)-axis in our reference frame so it coincides with the direction of travel. We are then able to write its position along the direction of linear motion as a simple function of time itself: \( x(t) = \nu \cdot t \). This combination of a pointlike charge zittering around some center and the oscillation as a whole them moving linearly is visualized below\(^{20}\): the radius

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\(^{16}\) We think the non-zero dimension of the elementary charge explains the small anomaly in the magnetic moment which should, therefore, not be thought as being anomalous. We must thank Prof. Dr. Randolf Pohl from the Max Planck Institute for Quantum Optics for, effectively, telling us to not think of the anomalous magnetic moment as an anomaly. For more details, see our paper on the electron model.

\(^{17}\) It is a derivation one can also use to derive a theoretical radius for the proton (or for any elementary particle, really). It works perfectly well for the muon, for example. However, for the proton, an additional assumption in regard to the proton’s angular momentum and magnetic moment is needed to ensure it fits the experimentally established radius. We shared the derivation with Prof. Dr. Randolf Pohl and the PRad team but we did not receive any substantial comments so far, except for the PRad spokesman (Prof. Dr. Ashot Gasparan) confirming the Standard Model does not have any explanation for the proton radius from first principles and, therefore, encouraging us to continue our theoretical research. In contrast, Prof. Dr. Randolf Pohl suggested the concise calculations come across as numerological only. We hope this paper might help to make him change his mind!

\(^{18}\) Prof. Dr. Patrick LeClair, Introduction to Modern Physics, Course Notes (PH253), 3 February 2019, p. 10. Also see our physical interpretation of Compton scattering in our previous paper(s).

\(^{19}\) In case the reader wonders what other ring current model might be on the current ‘market of ideas’, we are very much intrigued by electron models based on Dirac-Kerr-Newman geometries (A. Burinski, 2018, 2019 and 2020) because they may answer the very fundamental question he asked us when we first submitted our thoughts to Prof. Dr. Alexander Burinski: “I know many people who considered the electron as a toroidal photon and do it up to now. I also started from this model about 1969 and published an article in JETP in 1974 on it: ‘Microgeons with spin’. [However] There was [also] this key problem: what keeps [the pointlike charge] in its circular orbit?” (Email from Dr. Burinski to the author dated 22 December 2018)

\(^{20}\) We thank Prof. Dr. Giorgio Vassallo and his publisher to let us re-use this diagram. It originally appeared in an article by
of the circulatory motion must effectively diminish as the electron gains speed.\textsuperscript{21}

\textbf{Figure 1}: The Compton radius must decrease with increasing velocity

We must, of course, immediately urge the reader to imagine the plane of oscillation to rotate or oscillate itself in line with our interpretation of the angular momentum vector $\hbar$ as a vector rather than a simple scalar quantity. Also, when thinking of the electron moving linearly in an electromagnetic field, its precessional motion will yield a very different trajectory than the \textit{Archimedes screw} or \textit{helical motion} which is illustrated above. However, this does not much of an impact on the geometric interpretation of the \textit{de Broglie} wavelength which we advanced in previous paper(s) but which we will not repeat here.\textsuperscript{22}

To conclude our remarks on \textbf{Figure 1}, we just want to draw the attention of our reader to what happens when the classical (linear) velocity of the electron nears lightspeed: while its rotational motion and, therefore, its angular momentum does not vanish\textsuperscript{23}, we can see that the circumference of the oscillation – which is nothing but the (circular) Compton wavelength – turns into a \textit{linear} wavelength in the process!\textsuperscript{24} This rather remarkable geometric property relates the ring current model electron model with our photon model, which we will not talk about either here, however.\textsuperscript{25} All we want to do before getting into the meat of the matter of this paper – which is the subject of wave equations in atomic lattices – is to highlight the relativistic invariance of the argument of the wavefunction in this geometric interpretation of it.

We can, effectively, denote the position and time \textit{in the reference frame of the electron itself} by $x'$ and $t'$. Of course, the position of the electron particle in its own reference frame will be equal to $x'(t') = 0$ for all $t'$, and the position and time in the two reference frames will be related by Lorentz’s equations\textsuperscript{26}:

\begin{equation}
\end{equation}

\textbf{References:}

\textbf{Francesco Celani, Giorgio Vassallo and Antonino Di Tommaso} (Maxwell’s equations and Occam’s Razor, November 2017).

\textsuperscript{21} However, we urge the reader to imagine imagine the plane of oscillation to rotate or oscillate itself in line with our interpretation of the angular momentum vector $\hbar$ as a vector rather than a simple scalar quantity. Also, when thinking of the electron moving linearly in an electromagnetic field, we will also have precessional motion.

\textsuperscript{22} See earlier references.

\textsuperscript{23} Its radius does diminish, however, and actually goes to zero: this should be compatible with relativity theory. However, we did not yet have the time to work out the math in this regard.

\textsuperscript{24} We may, therefore, think of the Compton wavelength as a \textit{circular} wavelength: it is the length of a circumference rather than a linear feature!

\textsuperscript{25} The reason is the same: we do not want to repeat text from previous papers. If interested, the reader can look up our paper on Relativity, Light and Photons.

\textsuperscript{26} These are the Lorentz equations in their simplest form. We may refer the reader to any textbook here but, as usual, we like
\[
x' = \frac{x - vt}{\sqrt{1 - \frac{v^2}{c^2}}} = \frac{vt - vt}{\sqrt{1 - \frac{v^2}{c^2}}} = 0
\]
\[
t' = \frac{t - \frac{vx}{c^2}}{\sqrt{1 - \frac{v^2}{c^2}}}
\]

Hence, if we denote the energy and the momentum of the electron in our reference frame as \( E_v \) and \( p = \gamma m_0 v \), then the argument of the (elementary) wavefunction \( a \cdot e^{i \theta} \) can be re-written as follows:\(^{27}\):

\[
\theta = \frac{1}{\hbar} (E_v t - px) = \frac{1}{\hbar} \left( \frac{E_0}{\sqrt{1 - \frac{v^2}{c^2}}}t - \frac{E_0 v}{\sqrt{1 - \frac{v^2}{c^2}}}x \right) = \frac{1}{\hbar} E_0 \left( \frac{t}{\sqrt{1 - \frac{v^2}{c^2}}} - \frac{vx}{c^2 \sqrt{1 - \frac{v^2}{c^2}}} \right) = \frac{E_0}{\hbar} t'
\]

\( E_0 \) is, obviously, the rest energy and, because \( p' = 0 \) in the reference frame of the electron, the argument of the wavefunction effectively reduces to \( E_0 t'/\hbar \) in the reference frame of the electron itself.

The reader may also note that, besides proving that the argument of the wavefunction is relativistically invariant, we also demonstrated the relativistic invariance of the Planck-Einstein relation,\(^{28}\) in the process! This is why we feel that the argument of the wavefunction (and the wavefunction itself) is more real – in a physical sense – than the various wave equations (Schrödinger, Dirac, or Klein-Gordon) for which it is some solution.

Having said that, wave equations is what we want to talk about it here, so let us get on with it.

**Schrödinger’s wave equation in free space**

We quoted Dirac and his wave equation rather extensively above and, hence, the reader may wonder why we do not start with Dirac’s own wave equation. The reason is this: at the occasion of the 1948 Solvay Conference, Paul Dirac himself admits his wave equation may not do the trick. Indeed, we may usefully quote the following remarks, which he makes as he tries to challenge Robert Oppenheimer’s presentation of the use of perturbation theory in quantum mechanics:

“All the infinities that are continually bothering us arise when we use a perturbation method, when we try to expand the solution of the wave equation as a power series in the electron charge. Suppose we look at the equations without using a perturbation method, then there is no reason to believe that infinities would occur. The problem, to solve the equations without using perturbation methods, is of course very difficult mathematically, but it can be done in some simple cases. For example, for a single electron by itself one can work out very easily the solutions without using perturbation methods and one gets solutions without infinities. I think it is true also for several electrons, and probably it is true generally: we would not get infinities if

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\(^{27}\) One can use either the general \( E = mc^2 \) or – if we would want to make it look somewhat fancier – the \( pc = Ev/c \) relation. The reader can verify they amount to the same.

\(^{28}\) The relativistic invariance of the Planck-Einstein relation emerges from other problems, of course! The added value of the model here is the geometric interpretation of it: the Planck-Einstein relation models the integrity of the idea of a particle!
we solve the wave equations without using a perturbation method. However, if we look at the solutions which we obtain in this way, we meet another difficulty: namely we have the run-away electrons appearing. Most of the terms in our wave functions will correspond to electrons which are running away, in the sense we have discussed yesterday. They can, therefore, not correspond to anything physical. Thus nearly all the terms in the wave functions have to be discarded, according to present ideas. Only a small part of the wave function has a physical meaning.\(^\text{30}\)

In our interpretation of matter-particles, this small part of the wavefunction is, of course, the real electron, and it is the ring current or Zitterbewegung electron: it is the trivial solution that Schrödinger had found, and which Dirac himself mentioned in his Nobel Prize lecture from which we quoted above. The other part of the solution(s) are, effectively, bizarre oscillations which Dirac here refers to as ‘run-away electrons’.

With the benefit of hindsight, one wonders why Dirac did not see what we see now.\(^\text{31}\) In any case, we may come back to Dirac’s equation later. For the time being, we want to start where it all started, and so that’s Schrödinger’s equation. Feynman derives it – without the term for the electrostatic potential around a positively charged nucleus\(^\text{33}\):

\[
\frac{\partial \psi}{\partial t} = i \frac{\hbar}{2m_{\text{eff}}} \nabla^2 \psi = i \frac{\hbar}{m} \nabla^2 \psi
\]

What is \(m_{\text{eff}}\)? It is the concept of the effective mass of an electron which, in our ring current model, corresponds to the relativistic mass of the electric charge as it zitters around at lightspeed and so we can effectively substitute \(2m_{\text{eff}}\) for the mass of the electron \(m = m_e = 2m_{\text{eff}}.\)\(^\text{34}\)

---

\(^{29}\) This corresponds to wavefunctions dissipating away. As we noted, the problem is that the matter-particles they purport to describe obviously do not dissipate away.

\(^{30}\) This is our translation from French: Dirac must have made his remarks in his native language (English) but we were not able to find these. See pp. 282-283 of the report of the 1948 Solvay Conference, Discussion du rapport de Mr. Oppenheimer.

\(^{31}\) One of our correspondents wrote us this: “Remember these scientists did not have all that much to work with. Their experiments were imprecise – as measured by today’s standards – and tried to guess what is at work. Even my physics professor in 1979 believed Schrödinger’s equation yielded the exact solution (electron orbitals) for hydrogen.” Hence, perhaps we should not be surprised. However, in light of the intellectual caliber of these men, we are.

\(^{32}\) We probably Richard Feynman too much, but his hyperbole usually does reflect mainstream perceptions, so let us quote him once more: “Schrödinger’s equation was the first quantum-mechanical equation ever known. It was written down by Schrödinger before any of the other quantum equations we have described in this book were discovered. Although we have approached the subject along a completely different route [Feynman refers to the derivation of the Schrödinger’s equation wave equation in an atomic lattice here, so that is Schrödinger’s equation without the Coulomb term], the great historical moment marking the birth of the quantum mechanical description of matter occurred when Schrödinger first wrote down his equation in 1926.”

\(^{33}\) For Schrödinger’s equation in free space or the same equation with the Coulomb potential see Chapters 16 and 19 of Feynman’s Lectures on Quantum Mechanics respectively. Note that we moved the imaginary unit to the right-hand side, as a result of which the usual minus sign disappears: \(1/i = -i.\)

\(^{34}\) For a derivation of the \(m = 2m_{\text{eff}}\) formula, we refer the reader to our paper on the ring current model of an electron, where we write the effective mass as \(m_{\text{eff}} = m_e.\) The gamma symbol (\(\gamma\)) refers to the photon-like character of the charge as it zips around some center at lightspeed. However, unlike a photon, a charge carries charge. Photons do not. Hence, we agree – with Burinskii – that one should not use terms such as toroidal photon to refer to the pointlike Zitterbewegung charge even if its rest mass – just like that of a photon – is equal to zero, which is why it (also) whizzes around at lightspeed.
So far, so good. The question now is: are we talking one wave or many waves? A wave packet or the elementary wavefunction? Let us first make the analysis for one wave only, assuming that we can effectively write \( \psi \) as the \textit{elementary} wavefunction \( \psi = a \cdot e^{i \phi} = a \cdot e^{i(\lambda x - \omega t)} \).

Now, two complex numbers \( a + i \cdot b \) and \( c + i \cdot d \) are equal if, and only if, their real and imaginary parts are the same, and the \( \frac{\partial \psi}{\partial t} = i(\hbar/m) \cdot \nabla^2 \psi \) equation amounts to writing something like this: \( a + i \cdot b = i(c + i \cdot d) \). Remembering that \( \lambda^2 = -1 \), you can then easily figure out that \( i(c + i \cdot d) = i \cdot c + \lambda^2 \cdot d = -d + i \cdot c \). The \( \frac{\partial \psi}{\partial t} = i(\hbar/m) \cdot \nabla^2 \psi \) wave equation therefore corresponds to the following set of equations:\(^35\):

- \( \text{Re}(\frac{\partial \psi}{\partial t}) = -(\hbar/m) \cdot \text{Im}(\nabla^2 \psi) \iff \omega \cdot \cos(kx - \omega t) = k^2(\hbar/m) \cdot \cos(kx - \omega t) \)
- \( \text{Im}(\frac{\partial \psi}{\partial t}) = (\hbar/m) \cdot \text{Re}(\nabla^2 \psi) \iff \omega \cdot \sin(kx - \omega t) = k^2(\hbar/m) \cdot \sin(kx - \omega t) \)

It is, therefore, easy to see that \( \omega \) and \( k \) must be related through the following relation:\(^36\):

\[
\omega = \frac{\hbar k^2}{m} = \frac{\hbar c^2 k^2}{E}
\]

We can easily verify this makes perfect sense if we substitute the energy \( E \) using the Planck-Einstein relation \( E = \hbar \cdot \omega \) and assuming the wave velocity is equal to \( c \), which should be the case if we are talking about the same vacuum as the one through which Maxwell’s electromagnetic waves are supposed to be traveling:\(^37\):

\[
\omega = \frac{\hbar k^2}{m} = \frac{\hbar c^2 k^2}{E} = \frac{\hbar c^2 k^2}{\hbar \omega} = \frac{c^2 k^2}{\omega} \iff \frac{\omega^2}{k^2} = \frac{(2\pi f)^2}{(2\pi/\lambda)^2} = (f \lambda)^2 = c^2 \iff c = f \lambda
\]

We know need to think about the question we started out with: one wave or many component waves? It is fairly obvious that if we think of many component waves, each with their own frequency, then we need to think about different values \( m \) or \( E \) for the mass and/or energy of the electron as well! How can we motivate or justify this? The electron mass or energy is known, isn’t it? This is where the \textit{uncertainty} comes in: the electron may have some (classical) velocity or momentum for which we may not have a \textit{definite} value. If so, we may assume different values for its (kinetic) energy and/or its (linear) momentum may be possible. We then effectively get various \textit{possible} values for \( m \), \( E \), and \( p \) which we may denote as \( m_i \), \( E_i \), and \( p_i \), respectively. We can, then, effectively write our dispersion relation and, importantly, \textit{the condition for it to make physical sense} as:

\[
\omega_i = \frac{\hbar k^2}{m_i} = \frac{\hbar c^2 k_i^2}{E_i} = \frac{\hbar c^2 k_i^2}{\hbar \omega_i} = \frac{c^2 k_i^2}{\omega_i} \iff \frac{\omega_i^2}{k_i^2} = c^2 \iff c = f_i \lambda_i
\]

Of course, the \( c = f \lambda_i \) makes \textit{a lot} of sense: we would not want the properties of the medium in which

\(^35\) We invite the reader to double-check our calculations. If needed, we provide some more detail in one of our physics blog posts on the geometry of the wavefunction.

\(^36\) If you google this (check out the Wikipedia article on the dispersion relation, for example), you will find this relation is referred to as a \textit{non-relativistic} limit of a supposedly relativistically correct dispersion relation, and the various authors of such accounts will usually also add the 1/2 factor because they conveniently (but \textit{wrongly}) forget to distinguish between the \textit{effective} mass of the Zitterbewegung charge and the total energy or mass of the electron as a whole.

\(^37\) We apologize if this sounds slightly ironic but we are actually astonished Louis de Broglie does not mind having to assume superluminal speeds for wave velocities, even if it is for \textit{phase} rather than group velocities.
matter-particles move to be different from the medium through which electromagnetic waves are travelling: lightspeed should remain lightspeed, and waves – matter-waves included – should not be traveling faster. Let us quickly sum up our key conclusions so far:

1. If there is a matter-wave, then it must travel at the speed of light and not, as Louis de Broglie suggests, at some superluminal velocity.

2. If the matter-wave is a wave packet rather than a single wave with a precisely defined frequency and wavelength, then such wave packet will represent our limited knowledge about the momentum and/or the velocity of the electron. The uncertainty is, therefore, not inherent to Nature, but to our limited knowledge about the initial conditions.

Let us now look at Schrödinger’s wave equation in the context of a crystal lattice. This should be easy enough because Feynman actually derives Schrödinger’s wave equation in the very same context. He – and all academics who produced more recent textbooks based on his – then just substitutes the efficient mass \( m_{\text{eff}} \) for \( m_e \) rather than by \( m_e/2 \) noting, without any explanation at all, that the effective mass of an electron becomes the free-space mass of an electron outside of the lattice:

\[
\frac{\partial \psi}{\partial t} = i \frac{\hbar}{2m_{\text{eff}}} \nabla^2 \psi = i \frac{\hbar}{2m} \nabla^2 \psi
\]

In fact, it is a happy or not-so-happy coincidence that this 1/2 mistake is not noticed when adding the electrostatic potential term so as to model an atom because – as we know – the solutions to Schrödinger’s equation for electron orbitals are actually solutions for orbitals with two electrons, so that is what then take care of the \( \frac{1}{2} \) factor: the only reason why Schrödinger’s equation for an atom – with the electrostatic potential and the \( \frac{1}{2} \) factor works is because it models electron orbitals for two electrons. In fact, that the substitution of the effective mass \( m_{\text{eff}} \) by the total mass of the electron \( m_e \) is rather nonsensical also follows from Feynman’s casual remark in another chapter which uses the same concepts to arrive at the same equation:

"Don’t forget that \( m_{\text{eff}} \) has nothing to do with the real mass of an electron. It may be quite different—although in commonly used metals and semiconductors it often happens to turn out to be the same general order of magnitude, about 0.1 to 30 times the free-space mass of the electron."

In fact, we may usefully note here that, in the original 1963 print edition we bought long time ago, the general order of magnitude is actually mentioned as being equal to “about 2 to 20 times” the mass of an electron. We think we should understand all of the above as follows: the effective mass \( m_{\text{eff}} \) can be (almost) any ratio or multiple of the total mass \( m_e \) but \( m_{\text{eff}} \) being equal to \( m_e \), exactly, is the exception rather than the rule! To be precise, the exception is this: the ratio between the effective mass and the total mass is actually \( \frac{1}{2} \) in free space – so we should actually write Schrödinger’s equation in free space as \( \frac{\partial \psi}{\partial t} = i \frac{\hbar}{m} \nabla^2 \psi \). However, because the equation – in the context of electron orbitals – models the orbitals for two electrons, the \( \frac{1}{2} \) and 2 factor cancel each other out, and we get the Schrödinger equation we all memorized:

\[
38 \text{ Feynman’s Lectures, Vol. III, Chapter 13, Section 3, Time-dependent states.}
39 \text{ The equation is usually written using the } V(r) = -\frac{e^2}{r}, \text{ with } e \text{ the electron charge and } r \text{ the distance between the proton and}
\]
\[
\frac{\partial \psi}{\partial t} = i\left(\frac{\hbar}{2m} \nabla^2 + \frac{e^2}{\hbar r}\right)\psi
\]

There is another accident in the history of ideas which we must mention here—and one that is definitely unhappy: for some reason I, once again, cannot understand, Feynman follows de Broglie and earlier theorists in associating the same \( \frac{1}{2} \) factor in Schrödinger’s equation with the \( \frac{1}{2} \) factor in the non-relativistic \( KE = \frac{m v^2}{2} \) formula for kinetic energy and says we need a different wave equation, which is the Klein-Gordon equation. The Klein-Gordon relation is developed in pretty much the same way as Schrödinger’s equation: a heuristic argument is used but this time it is the relativistically correct energy-momentum relation which, combining the \( E = mc^2 \) (Planck-Einstein) and \( p = \frac{h}{\lambda} \) (de Broglie) relations, serves as the dispersion relation—as opposed to the much more intuitive \( \omega = \frac{\hbar k^2}{m} \) relation. We will come back to this later.

**Schrödinger’s wave equation in a crystal lattice**

**Feynman’s lattice**

Feynman writes the following about his quantum-mechanical derivation of Schrödinger’s equation in crystal lattice:

"We do not intend to have you think we have derived the Schrödinger equation but only wish to show you one way of thinking about it. When Schrödinger first wrote it down, he gave a kind of derivation based on some heuristic arguments and some brilliant intuitive guesses. Some of the arguments he used were even false, but that does not matter; the only important thing is that the ultimate equation gives a correct description of nature.\(^{40}\)

Unfortunately, after playing with it for a while, he then discards it for the wrong reason:

"In principle, Schrödinger’s equation is capable of explaining all atomic phenomena except those involving magnetism and relativity. [...] The Schrödinger equation as we have written it does not take into account any magnetic effects. It is possible to take such effects into account in an approximate way by adding some more terms to the equation. However, as we have seen in Volume II, magnetism is essentially a relativistic effect, and so a correct description of the motion of an electron in an arbitrary electromagnetic field can only be discussed in a proper relativistic equation. The correct relativistic equation for the motion of an electron was discovered by Dirac a year after Schrödinger brought forth his equation, and takes on quite a different form. We will not be able to discuss it at all here."\(^{41}\)

However, as we note above, Dirac’s equation is actually *not* correct (its solutions are dissipating wave packets: ‘run-away electrons’), and the \( \frac{1}{2} \) factor in Schrödinger’s equation may be relativistically correct the (two) electron(s). These remarks should allow the reader to relate our formulation to the rendering he or she is used to.

\(^{40}\) Lectures, Vol. III, Chapter 16, p. 16-4.

\(^{41}\) Lectures, Vol. III, Chapter 16, p. 16-13. We have re-read Feynman’s Lectures many times now and, in discussions with fellow amateur physicists, we sometimes joke that Feynman must have had a secret copy of the truth. He clearly doesn’t bother to develop Dirac’s equation because – having worked with Robert Oppenheimer on the Manhattan project – he knew Dirac’s equation only produces non-sensical ‘run-away electrons’. In contrast, while noting Schrödinger’s equation is non-relativistic, it is the only one he bothers to explore extensively. Indeed, while claiming the Klein-Gordon equation is the ‘right one’, he hardly devotes any space to it.
because it is related to the $\frac{1}{2}$ factor in the $m_{\text{eff}} = m_e/2$ which relates the effective mass to the total mass in the ring current model of an electron! In fact, we may quickly insert the Klein-Gordon wave equation here which, according to Feynman, corrects Schrödinger’s supposedly non-relativistically correct wave equation:

$$\frac{1}{c^2} \frac{\partial^2 \psi}{\partial t^2} - \frac{m^2 c^2}{\hbar^2} \psi = \nabla^2 \psi$$

As explained above, we have no use for it because its solutions – any quantum-mechanical wave packet that comes out of it as a solution – dissipate away. Actual matter-particles do not. We do not adhere to mainstream wisdom here, according to which this equation incorporates a relativistically correct dispersion relation: superluminal wave velocities – even phase velocities – are effectively not compatible with relativity theory! Let us, therefore, explore Feynman’s derivation of Schrödinger’s equation – the very first wave equation in the history of quantum mechanics! – in all of its detail.

In case you wonder, we do not do this to show off Feynman’s brilliance (we assume no one needs any convincing in this regard) but to get a better understanding of what might be real and what isn’t and – even more importantly – to better understand the difference between a matter-wave, a signal, an amplitude and the electron itself, as per the title of this paper.

You will, of course, understand our analysis will be as heuristic as Feynman’s because we have a clear objective in mind: we want to show the moving electron is the moving electron, and the amplitude is the signal. Hence, we request the reader to be very critical but constructive: when everything is said and done, a probability must represent something, and if it isn’t the electron, then it should be the signal.

What other common-sense candidates for reality are there? Let us, so as to focus our mind on what is what, copy Feynman’s illustration of – paraphrasing Wittgenstein here – whatever it is that might be the case (Figure 2).

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42 See: Richard Feynman, Waves in three dimensions, *Lectures, Vol. I, Chapter 48*. With the usual hyperbole, Feynman refers to it as the ‘grand equation, which corresponds to the dispersion equation for quantum-mechanical waves.’ In fact, because his students are – at that point – not yet familiar with differential calculus for vector fields (and, therefore, not with the Laplacian operator $\nabla^2$), Feynman writes it there like this:

$$\frac{\partial^2 \phi}{\partial x^2} + \frac{\partial^2 \phi}{\partial y^2} + \frac{\partial^2 \phi}{\partial z^2} - \frac{1}{c^2} \frac{\partial^2 \phi}{\partial t^2} = \frac{m^2 c^2}{\hbar^2} \phi$$

We are not ashamed to admit Feynman’s early introduction of this equation in his three volumes of lectures on physics which, as he clearly states in his preface, were written “to maintain the interest [in physics] of the very enthusiastic and rather smart students coming out of the high schools” did not miss their effect on us: I wrote this equation on a piece of paper on the backside of my toilet of my student room when getting my first degree (in economics) and vowed that, one day, I would understand this equation “the way I would like to understand it.” We now understand it to not represent anything real: I now understand it to model our uncertainty (our lack of knowledge about the past or initial conditions) rather than any fundamental uncertainty in Nature!

43 The first proposition of Wittgenstein’s *Tractatus Logico-Philosophicus* (originally published in his native German language as the *Logisch-philosophische Abhandlung*) is this: “Die Welt ist alles, was der Fall ist” (“The world is all that is the case”) and it is usually interpreted as Wittgenstein’s definition of reality. One has to start somewhere, isn’t it?
Figure 2: Feynman’s idea of a moving electron in an atomic lattice

We should note, first, that this is a very classical picture: Feynman talks quantum mechanics but, to focus the mind, he pictures an electron as always being somewhere, and this somewhere is near an atom. In Feynman’s words:

“We want to see what happens if we put a single electron on this line of atoms. Of course, in a real crystal there are already millions of electrons. But most of them (nearly all for an insulating crystal) take up positions in some pattern of motion each around its own atom—and everything is quite stationary. However, we now want to think about what happens if we put an extra electron in. We will not consider what the other ones are doing because we suppose that to change their motion involves a lot of excitation energy. We are going to add an electron as if to produce one slightly bound negative ion. In watching what the one extra electron does we are making an approximation which disregards the mechanics of the inside workings of the atoms. Of course the electron could then move to another atom, transferring the negative ion to another place. We will suppose that just as in the case of an electron jumping between two protons, the electron can jump from one atom to the neighbor on either side with a certain amplitude.”

At this point, we may already want to distinguish between the motion of the electron, and the possible motion of a whole series of electrons, as illustrated in Figure 3. This illustration is probably not by best drawing ever but you get the idea: think of the musical chairs game but with the chairs on a line and all players agreeing to kindly move to the next chair for the new arrival and – importantly – the last person on the last chair agreeing to leave the game to get a beer.

Figure 3: The idea of a moving electric signal in an atomic lattice

What happens in such game of musical chairs is the transmission of a signal: while each person needs

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We will (mostly) quote from Chapter 13 (Propagation in a Crystal Lattice) as well as Chapter 16 (Dependence of Amplitudes on Position) in Feynman’s Lectures on Quantum Mechanics here. Chapter 13 focuses on the wavefunction, whereas Chapter 16 focuses more on the wave equation that might generate them as solutions.
some time to get up – say $\Delta t$ – and move to the next chair, an orchestra leader could, perhaps, make them all move at the same time and, hence, while each person would move only one chair, the transmission of the signal could be lightning fast. You may wonder: what is the signal here? It is just this: a new person has arrived and took the first chair.

The difference between a traveling electric charge and a traveling electric signal
What we explained above is, of course, the idea of an electron current: we know the typical drift speed of a free electron is actually quite low – it collides with many other charged particles (electron shells or other free electrons) – but the electric signal travels much faster. Why? When charged particles are forced into the wire – or any conductor really – an equal number are forced to leave because of the repulsion between like charges. That makes it actually difficult to increase the number of charges in a volume. Even if one charge enters, another leaves almost immediately, carrying the signal rapidly forward.\textsuperscript{45}

The question is how fast, exactly? We checked but find that textbooks will agree that the order of magnitude is the same as that of light but seem to shy away from actually calculating signal speeds. One course compares signal and charge drift velocity as follows: “Electrical signals are known to move very rapidly. Telephone conversations carried by currents in wires cover large distances without noticeable delays. Lights come on as soon as a switch is flicked. Most electrical signals carried by currents travel at speeds on the order of $10^8$ m/s, a significant fraction of the speed of light. Interestingly, the individual charges that make up the current move much more slowly on average, typically drifting at speeds on the order of $10^{-4}$ m/s.”\textsuperscript{46}

We may rephrase the question: if the velocity of an electromagnetic field in free space is equal to $c$, then at what fraction of $c$ can electric signals travel in a conductor, a semiconductor or whatever lattice-like structure? Lightspeed itself? Slower? Faster? Let us get back to Feynman’s analysis of quantum-mechanical amplitudes and the associated wave equation(s) to see whether or not it may help to answer that question. The modeling is this\textsuperscript{47}:

1. The assumption is that the amplitude for an electron to go from one atom to the next is given by $iA/\hbar$ per unit time. Such assumption must, of course, assume that the electron that goes from one atom to the next starts out at the first atom which must, therefore, have a different energy. We, therefore, know – from the analysis of two-state systems in quantum mechanics – that $A$ must be related to some energy difference.\textsuperscript{48}

Feynman does not give any explanation here but we may want to think, for example, of the extra

\textsuperscript{45} It is quite astonishing but Feynman actually makes the case that a current-carrying wire is actually neutral: it only appears charged when the charges inside are set in motion. See Feynman’s lecture on the relativity of magnetic and electric fields.

\textsuperscript{46} We simply googled and this course (from a private course provider which we must reference and/or credit here) was just one of the many textbooks to pop up.

\textsuperscript{47} We stick closely to Feynman’s presentation here but, at the same time, we want to shorten and we do have our own nuances in the interpretation, so the reader should check with the original. We will try to limit our additions and critical remarks to footnotes.

\textsuperscript{48} However, the reader should this system is not quite the same as a simple two-state system, in which we the two energy levels can be written as $E_0 + A$ and $E_0 - A$ states. The difference between the two energy levels in a two-state system is, therefore, $2A$, while the difference between these two states and the zero state $E_0$ is equal to $A$. 


electron at atom \( n \) as occupying some orbital and its angular momentum will, therefore, be equal to some multiple of \( \hbar \). Such visualization or modeling has the advantage that it allows us to effectively interpret \( A/\hbar \) as some energy per unit time.\(^{49}\)

The reader should also note that the amplitude is constant: it, therefore, increases linearly with time. Likewise, the associated probability (the squared absolute value) will also not increase exponentially but linearly.\(^{50}\) So far so good.

One obvious question might be this: do we allow the electron to go back? If \( iA/\hbar \) is the amplitude to go from atom \( n \) to atom \( n+1 \), then its (complex) conjugate, \( -iA/\hbar \), should be the amplitude to go from atom \( n \) to atom \( n-1 \), isn't it? That sounds logical but that would amount to time reversal, which we do not allow: time goes in one direction only.\(^{51}\) That is why we have the imaginary unit in our \( iA/\hbar \) expression: it comes with the chosen convention for modeling wavefunctions.

All of this looks very reasonable. Hence, let us move on to the next assumption.

2. The electron must be somewhere when it goes from one atom to another. Feynman writes the amplitude to on be on the \( n^{th} \) atom as \( C_n \) and suggests to the following general functional form for it:

\[
C_n = a_n e^{-\frac{E_n}{\hbar} t} = a(x_n) e^{-\frac{E_n}{\hbar} t} = a(n \cdot b) e^{-\frac{E_n}{\hbar} t}
\]

This looks pretty good — the amplitude varies with time so as to capture our uncertainty — but why the subscript in the \( a_n \) coefficient? Should the probability (the absolute square of the amplitude) not be the same for every atom? No. If the electron is not anywhere near atom \( n \), it will probably not get there in the next instant. Hence, these amplitudes depend one on another.\(^{52}\)

Next assumption.

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\(^{49}\) See Chapter VII (The wavefunction and the atom) of our book on a realist interpretation of quantum mechanics, in which we assume the \( n^{th} \) orbital (this \( n \) has, obviously, nothing to do with the atom number \( n \) in our lattice) to pack an angular momentum that is equal to \( n \cdot \hbar \). It is then easy to show — at least for a simple hydrogen atom — that this orbital will be associated with an energy that is equal to \( E_n = -\frac{1}{n^2} E_R \) and an angular momentum that will be equal to \( L_n = L \cdot \omega_n = nh \). The Planck-Einstein relation — written as \( E_n = n \cdot h \cdot \omega_n \) when modeling Bohr orbitals — then tells us the (angular) period or cycle time will be equal to \( T_n = \frac{1}{2 \pi f} = \frac{1}{\omega_n} = \frac{\hbar}{E_n} \). We can, therefore, see that we may now think of equating \( A/\hbar \) to \( E_n/\hbar \cdot h \Rightarrow A = E_n/n \) and that \( \frac{A}{\hbar} = \frac{E_n}{nh} = \frac{2\pi}{T_n} = \omega_n \) can effectively be interpreted as an energy per unit time. This unit time corresponds logically to an angular period.

\(^{50}\) This amounts to saying the past has no influence on the future here: it is not because an electron has been sitting on top of an atom for a while now that there is a greater likelihood it will soon make the next hop.

\(^{51}\) In our paper on classical interpretations of quantum-mechanical concepts we argue that the convention that is used in regard to the plus or minus sign for the imaginary unit in the \( ae^{-i\theta} = ae^{i(\theta+\pi)} \) also fixes the direction of time. Such convention is necessary because time goes in one direction only. Hence, when taking the complex conjugate of an wavefunction or an amplitude you perform a mathematical operation that amounts to time reversal but this does not correspond to a real possibility in physics: time must go in one direction only. In the mentioned paper we show this is essentially rooted in the idea of motion itself: if we would allow for time reversal, we would not be able to define well-defined equations of motion.

\(^{52}\) As for the substitution of \( a_n \), one can easily appreciate \( a_n \) is a function of \( n \) or, what amounts to the same, the position \( x_n = n \cdot b \). The latter substitution assumes we choose the origin of our line at atom zero and then uses the spacing between atoms \( b \) to get the rather logical \( x_n = n \cdot b \) equation.
3. We already know we want a wavefunction for our electron. Hence, remembering a wavefunction usually looks like $e^{-i(\omega t - kx)}$, Feynman suggests the following trial solution for $a_n = a(x_n)$:

$$a(x_n) = e^{ikx_n}$$

This gives us the wavefunction we had expected to see all along:\n
$$C_n = e^{ikx_n}e^{-\frac{E_n}{\hbar}} = e^{-i(\omega t - kx_n)}$$

Richard Feynman shows off by showing how one can associate a set of Hamiltonian equations with this system but we conveniently look at this as unnecessary luggage for the time being. The question is this: we have some idea of what the $\omega$ frequency actually is, but what is $k$? It is an important question because it obviously answers one of the basic questions we started out with: what is the propagation speed of this wave?

We have the model now. Hence, let us try to tackle all interpretational issues in the next section—if only to keep this rather long paper somewhat organized. Before we do so, we would like to note one thing here: Feynman gets a single wave. No wave packet. No uncertainty. Not in Nature, at least: once again, the uncertainty is only in our mind. 😊

The energy of an electron in a crystal lattice

Feynman starts by noting $k$ must be a real number: if it were an imaginary number, say $i\cdot k'$, we’d get a real exponential function for the coefficient ($a_n = e^{-ik}$) and depending on your convention for the $\pm$ sign for the imaginary unit for your wavefunction (or the $\pm$ sign for $k'$), the amplitudes would blow up or, else, die off. In addition, we have no coefficient in front of our $C_n = e^{-i(\omega t - kx)}$ wavefunction and its real and imaginary part, therefore, oscillate between $-1$ and $+1$. What more can we say about $k$? We should be able to relate it to the energy $E$, isn’t it?

Correct, and that is where the model becomes more complicated. We joked about Feynman showing off his mathematical skills by developing a set of Hamiltonian equations but we actually need them. The amplitude to be in this or that state depends on the amplitude to be in that or this state. For a simple two-state system, we have two energy levels that are separated from the zero state $E_0$ by $A$. These energy levels can, therefore, be written as $E_0 + A$ and $E_0 - A$ and the energy difference between them is, therefore, equal to $2A$. The Hamiltonian equations for such two-state system are these:\n
$$i\hbar \frac{dC_1}{dt} = E_0 C_1 - AC_2$$
$$i\hbar \frac{dC_2}{dt} = E_0 C_2 - AC_1$$

We know this modeling—with the imaginary unit ($i$) in these differential equations—leads to what we want to see: probabilities “sloshing back and forth” between the two states—an oscillatory solution.

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53 Richard Feynman shows off by showing how one can associate a set of Hamiltonian equations with this system but we conveniently look at this as unnecessary luggage.

54 See, for example, Feynman’s development of the set of Hamiltonian equations for an ammonia maser in Chapter 8 (The Hamiltonian matrix) of his Lectures.
rather than, say, what happens “when water leaks between two tanks”, which will simply result in the water levels approaching each other (the comparison is Feynman’s, of course). The reader will probably be familiar with the comparison with paired pendulums.

For this system – an electron moving through this linear array of atoms – we have a very large number of equations for a very large number of atoms. These equations look like this:

\[ i\hbar \frac{dC_n(t)}{dt} = E_0 C_n(t) - AC_{n+1}(t) - AC_{n-1}(t) \]

We refer the reader to Feynman’s rather ingenious solution to this potentially infinite set of equations (we may, perhaps, not imagine an actually infinite array but the reader will understand there are a lot of atoms in a lattice). One of the grand results is the formula for the energy \( E \), which is this:

\[ E = E_0 - 2A \cos(kb) \]

Figure 4 shows a graph of \( E \) as a function of \( k \): \( E - E_0 \) will be equal to \( 2A \) if \( \cos(kb) \) is equal to \(-1 \) and, hence, if \( kb = \pm \pi \iff k = \pm \pi b \). The reader can also verify the \( E - E_0 = -2A \) condition, which is fulfilled if \( k = 0 \).

![Figure 4: Energy E as a function of the parameter k](image)

It is a very particular graph, but it will be familiar to most engineers: an electron in a crystal will always have an energy within this \([E_0 - 2A, E_0 + 2A]\) band or, if we choose \( E_0 \) such that it is equal to \( 2A \), the \([0, 2A]\) band. No other energy is possible.

Let us now look at what we wanted to look at: wave velocities. Let us, to simplify things, effectively choose \( E_0 \) to be equal to \( 2A \), so our energy equation can be re-written as:

\[ E = 2A - 2A \cos kb = 2A(1 - \cos kb) \]

One single matter-wave or a wave packet?

We may now, once again, ask this question: should we think of a single wave or of a superposition of waves? Richard Feynman definitely thinks the motion of an electron should be represented by a wave packet and effectively calculates a group velocity using the familiar \( v_{\text{group}} = \frac{d\omega}{dk} \) equation. Let us go along with the argument for a while to see what does and does not make sense. First, we need to solve another small mathematical problem: how do you calculate \( \frac{d\omega}{dk} \) from the \( E = 2A(1 - \cos kb) \) equation?
Here, Feynman assumes we may use the small-angle approximation because, because of the very small space inbetween atoms, he assumes the amplitudes should only vary very little from one \( x_n \) to the next, which amounts to saying \( k \) must be very small.\(^{55}\) In such case, we may, perhaps, use the small-angle approximation for the cosine \((\cos \theta = 1 - \theta^2)\) and, therefore, write \( E \) as:

\[
E = 2A(1 - \cos kb) = 2A \left( 1 - 1 + \frac{k^2 b^2}{2} \right) = Ak^2 b^2
\]

Using the \( \omega = E/\hbar \) equation, we can now calculate wave velocities. Let us first calculate the phase velocity for a single wave:

\[
\nu_{\text{phase}} = \frac{\omega}{k} = \frac{Ak^2 b}{\hbar k} = \frac{Akb}{\hbar}
\]

What can we do with this? Nothing much. We would need to relate \( A \), somehow, to the energy of the electron or, to be precise, the difference in energies between an atom that has the extra electron and one that has not.\(^{56}\) We would also need to find some physical interpretation of \( k \). At this point, Feynman defines the concept of the effective mass of an electron. However, we do not want to use this because:

1. It has no relation whatsoever with the concept of effective mass as used in the context of the ring current model, in which it is associated with the relativistic mass of the pointlike charge as it zitters around at lightspeed.
2. Feynman himself admits his concept of “effective mass has nothing to do with the real mass of an electron. It may be quite different—although in commonly used metals and semiconductors it often happens to turn out to be the same general order of magnitude, about 0.1 to 30 times the free-space mass of the electron.”

What about the group velocity? We are not sure. We see no reason whatsoever to do what Feynman does, and that is to “make up a wave packet with a predominant wave number \( k_0 \), but with various other wave numbers near \( k_0 \).” We can then, effectively, use the \( E = Ak^2 b^2 \) as a dispersion relation and, therefore, calculate a group velocity which is equal to:

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\(^{55}\) That is a matter of scale, of course, and we, therefore, regret Feynman does not do a better job at motivating this assumption. In fact, it is worse than that: Feynman actually shows that, in any case, we may always replace a large \( k \) by a much smaller \( k \) by choosing a ratio between these two values that ensures the physical states that are being described are the same. The principle is illustrated below and we refer the reader to Feynman’s explanation of it, which is easy enough but, in our not-so-humble view, totally unacceptable! Indeed, if \( k \) is related to some real variable – which Feynman attempts to demonstrate – such as, for example, the momentum or velocity of the charge, then such substitutions should not be done, of course!

\(^{56}\) See our remarks in footnote 49.
\[
v_{\text{group}} = \frac{d\omega}{dk} = \frac{d(Ak^2b)}{dk} = \frac{2Ab^2}{\hbar}k
\]

And, yes, we can then define a totally artificial concept of effective mass by defining it as \( m_{\text{eff}} = \frac{\hbar^2}{2Ab^2} \) to write what Feynman apparently wants to write in some desperate effort to show the concept of a wave packet related to some kind of momentum concept might make sense:

\[
\hbar k = m_{\text{eff}} v_{\text{group}} = \frac{\hbar^2}{2Ab^2} \cdot \frac{2Ab^2}{\hbar}k = \hbar k
\]

Frankly, we do not see any use for it because of the above-mentioned reasons. We are not sure what use Feynman sees for it because he wraps up by summarizing what we have learned here as follows:

“\text{We have now explained a remarkable mystery—how an electron in a crystal (like an extra electron put into germanium) can ride right through the crystal and flow perfectly freely even though it has to hit all the atoms. It does so by having its amplitudes going pip-pip-pip from one atom to the next, working its way through the crystal. That is how a solid can conduct electricity.”}"

Frankly, we knew that already, didn’t we? In fact, Feynman is plain wrong here: an electron does not “ride right through the crystal”, nor does it “flow perfectly freely even though it has to hit all the atoms.” What does go “pip-pip-pip” from one atom to the next, is the signal and, yes, that is what these mysterious probability amplitudes actually model and – another yes – that is how a solid can conduct electricity. In short, we do not see any added value of Feynman’s quasi-infinite set of Hamiltonian equations. Why not? The whole model, and all of the machinery that comes with it, does not give any answer to the two basic questions:

1. What is the velocity of the electron itself?
2. What is the velocity of the signal?

Let us, therefore, wrap up this paper.
Conclusions
This paper basically further explored intuitions we highlighted in previous papers already:

1. The concept of the matter-wave traveling through the vacuum, an atomic lattice or any medium can be equated to the concept of an electric or electromagnetic signal traveling through the same medium.

2. There is no need to model the matter-wave as a wave packet: a single wave – with a precise frequency and a precise wavelength – will do.

3. If we do want to model the matter-wave as a wave packet rather than a single wave with a precisely defined frequency and wavelength, then the uncertainty in such wave packet reflects our own limited knowledge about the momentum and/or the velocity of the particle that we think we are representing. The uncertainty is, therefore, not inherent to Nature, but to our limited knowledge about the initial conditions.

4. The fact that such wave packets usually dissipate very rapidly, reflects that even our limited knowledge about initial conditions tends to become equally rapidly irrelevant. Indeed, as Feynman correctly notes, “the tiniest irregularities” tend to get magnified very quickly at the micro-scale.

In short, as Hendrik Antoon Lorentz noted a few months before his demise, there is no reason whatsoever “to elevate indeterminism to a philosophical principle.” We hope the added detail in this paper may finally convince our more skeptical readers.

Jean Louis Van Belle, 12 May 2020